## Hydrogène Orange: expériences et perspectives F. Osselin





## Natural Hydrogen

- Abiotic oxidation of iron:  $2FeO + H_2O = Fe_2O_3 + H_2$
- Radiolysis
- Magmatic degassing



PERIDOTITE

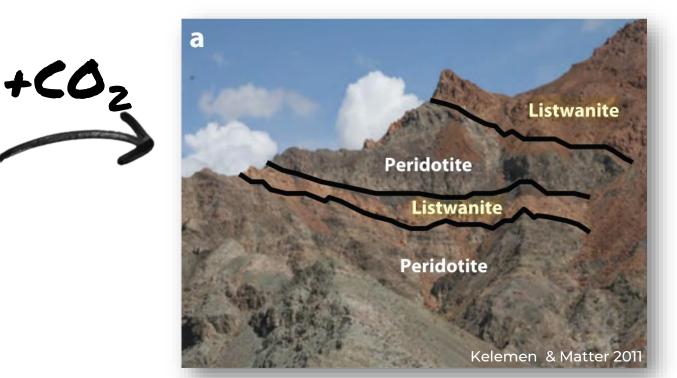
- 20Mt escaping every year from the subsurface to the atmosphere (Zgonnik 2020)
- 10<sup>20</sup> kg of peridotite on the first 7km of Earth (Kelemen 2008)
  - ✓ 1 kg<sub>H2</sub>/m<sup>3</sup>, potential 10<sup>8</sup>Mt

## **Natural Carbonation**

### Peridotite



# **Listwanite:** geological formation resulting from the extensive carbonation of peridotites



### 99,9% of total C stored as carbonates

 $(CAO, MGO)+CO_2 = (CA, MG)CO_3$ 

# **Orange Hydrogen: Stimulated Natural H<sub>2</sub>**

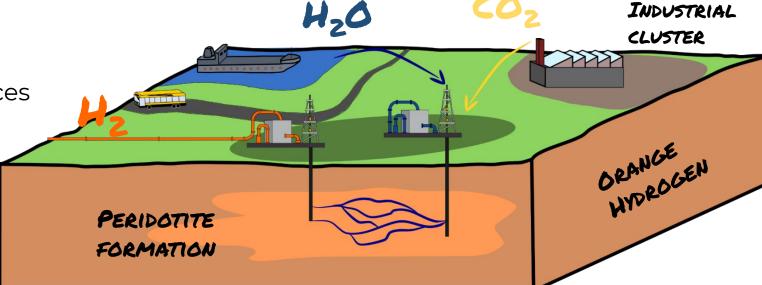
#### Stimulated production

 $\checkmark$  100 000 years of H<sub>2</sub>

- Controlled production rate
- ✓ Simplified exploration
- ✓ No excavation/mining
- ✓ No stress on freshwater resources

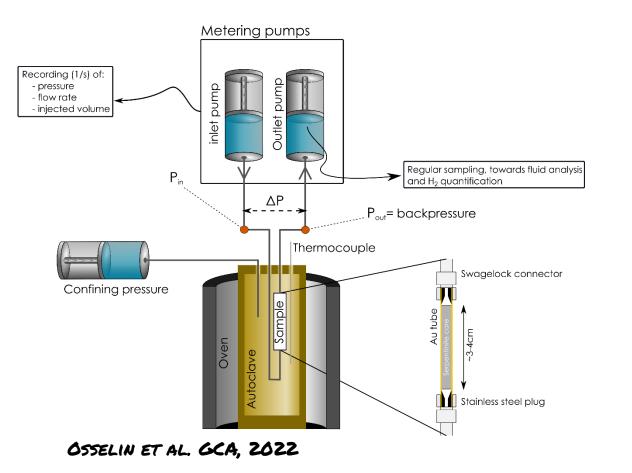
#### • Combination of CO<sub>2</sub> and H<sub>2</sub>

Economically more robust



- ★ More complicated process
- Requires injection & fracking
- ✗ Not « renewable »

# **Orange Hydrogen in the lab**



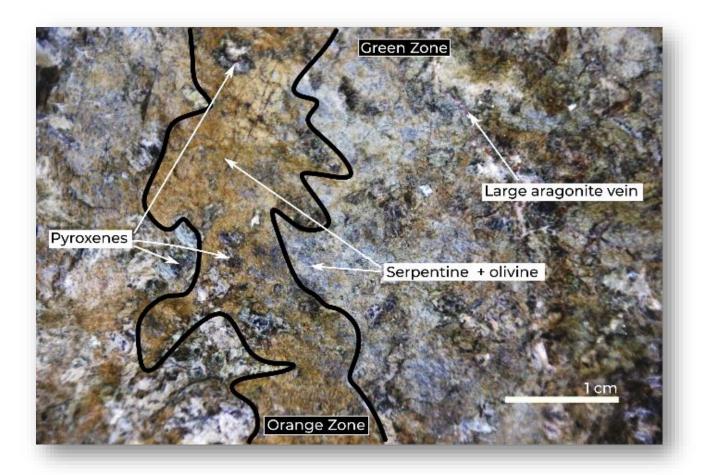
#### **Reactive percolation experiments**

Cores 5.6mm dia. – few cm long P < 500 bar – < 400°C





# Orange Hydrogen in the lab



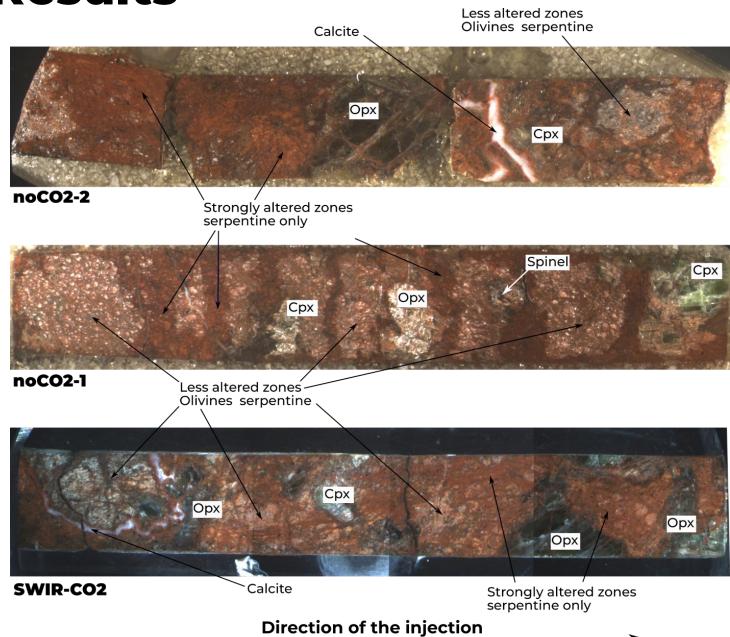
#### **Natural Serpentinite:**

50% Serpentine 20% Opx 11% Cpx 13% Olivine 5% Aragonite 1% Spinel

#### Injection at 280°C

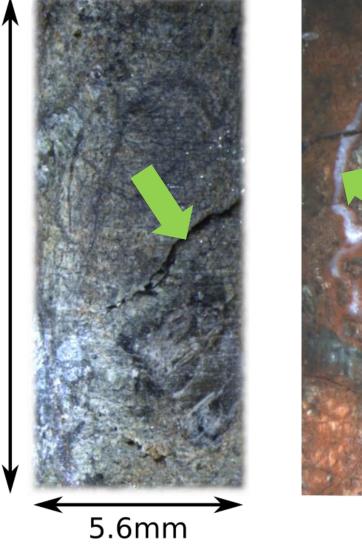
2 experiments with pure NaCl 1 experiment with NaCl+NaHCO<sub>3</sub>

## Results



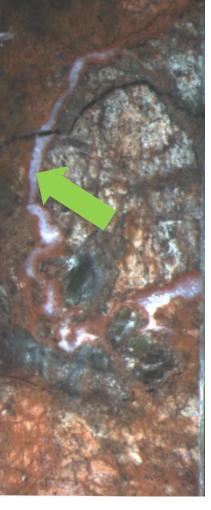
#### General decrease of carbonates

#### General increase of carbonates



СЗ

1.3

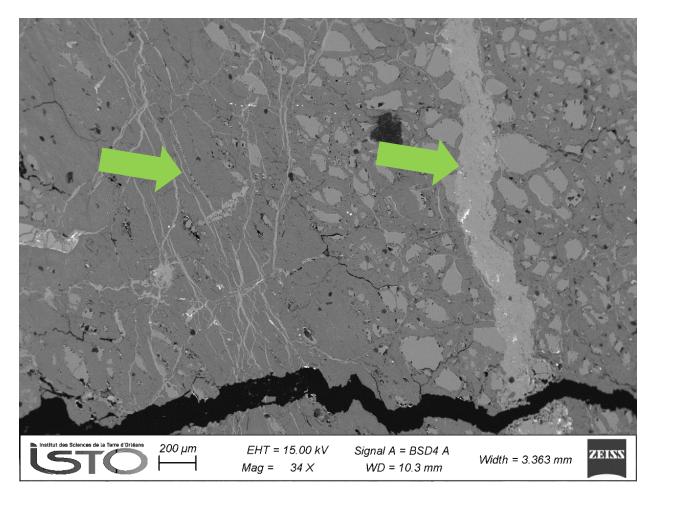


## 

Initially opened fractures get filled with calcite

BEFORE EXPERIMENT

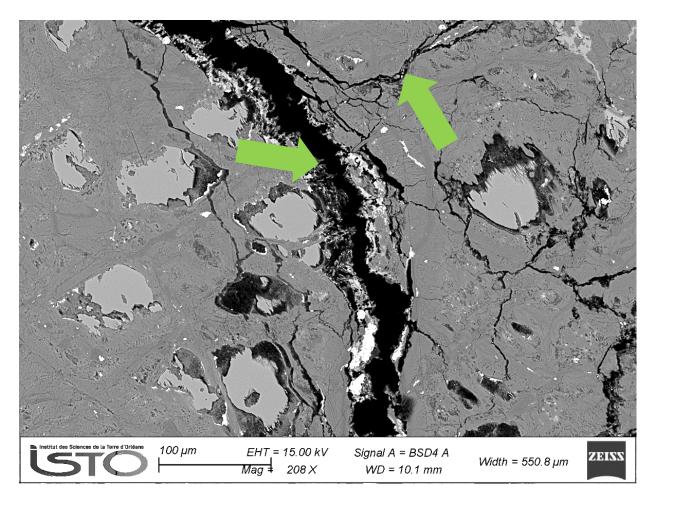
AFTER EXPERIMENT



## **CO**<sub>2</sub>

Initially opened fractures get filled with calcite

Secondary percolation paths also appear clogged with calcite (SEM) **Multiscale precipitation** 



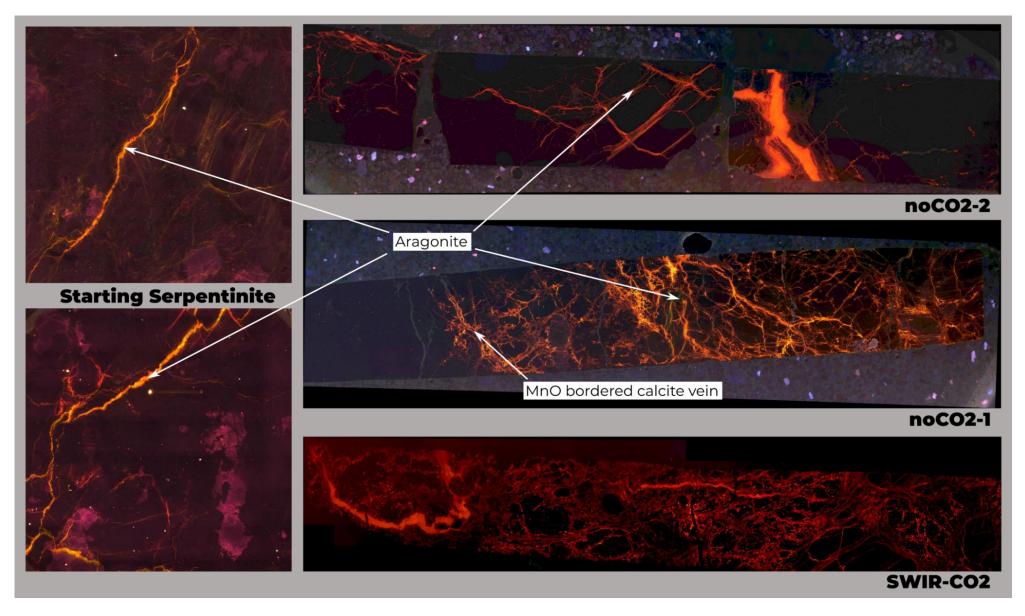
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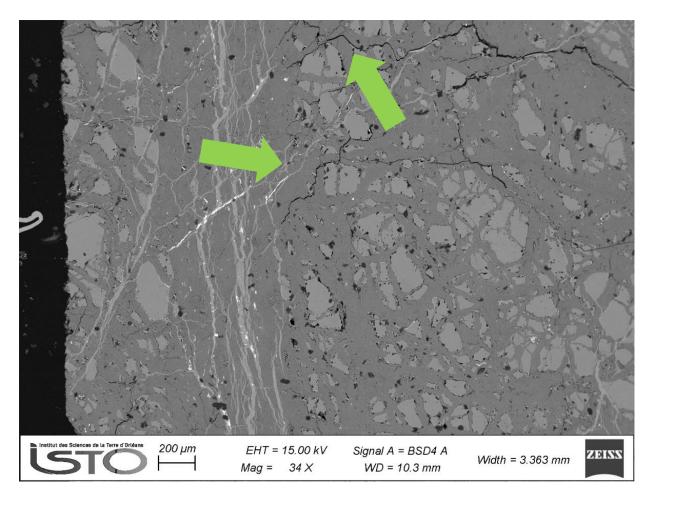
## 

Aragonite veins get emptied on all scales



MUCH DENSER AND BRIGHTER NETWORK OF CARBONATES - MNO POLLUTION FROM INOX CORROSION

## **Results** – Silicate alteration

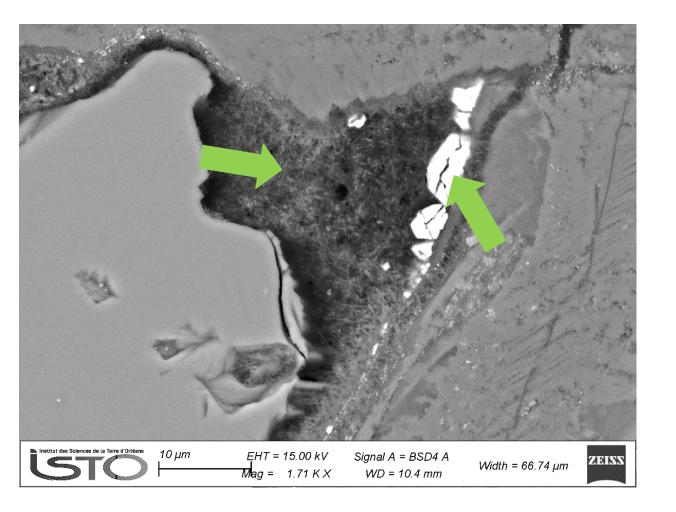


## 

Complete replacement of olivines by serpentine mesh in main and secondary percolations paths

Absence of olivine alteration in low flow zones

## **Results** – Silicate alteration



## 

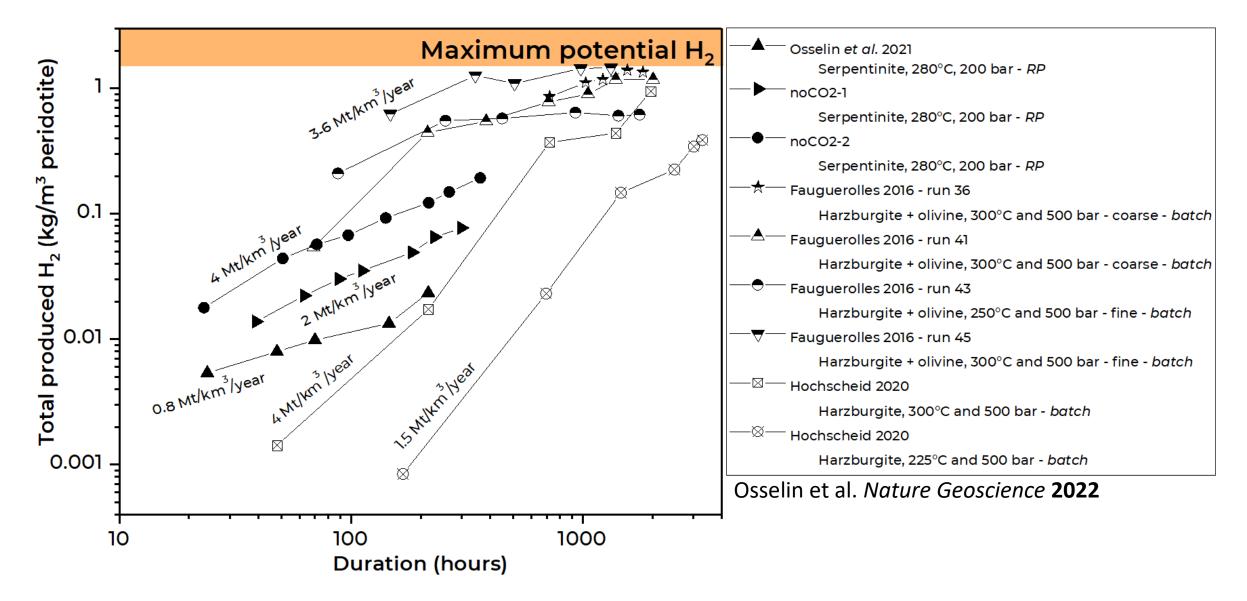
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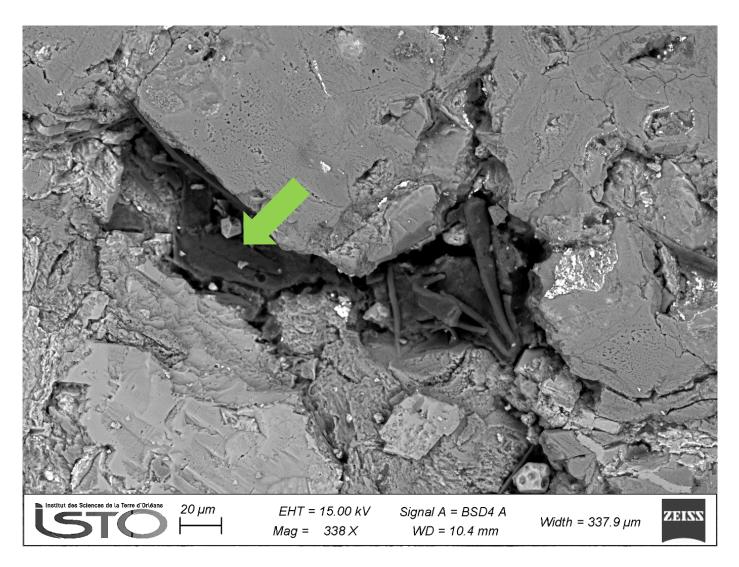
### noCO<sub>2</sub>

Similar replacement pattern but chrysotile instead of lizardite

# **Hydrogen production**



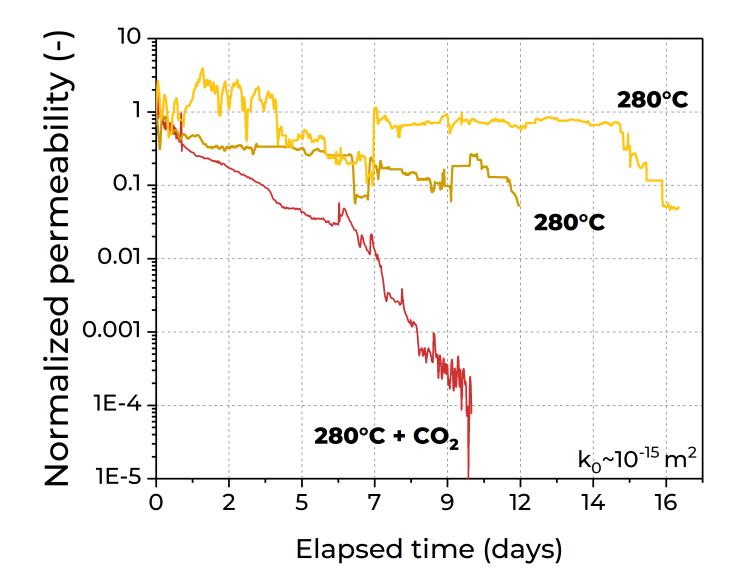
# $CO_2/H_2$ interaction



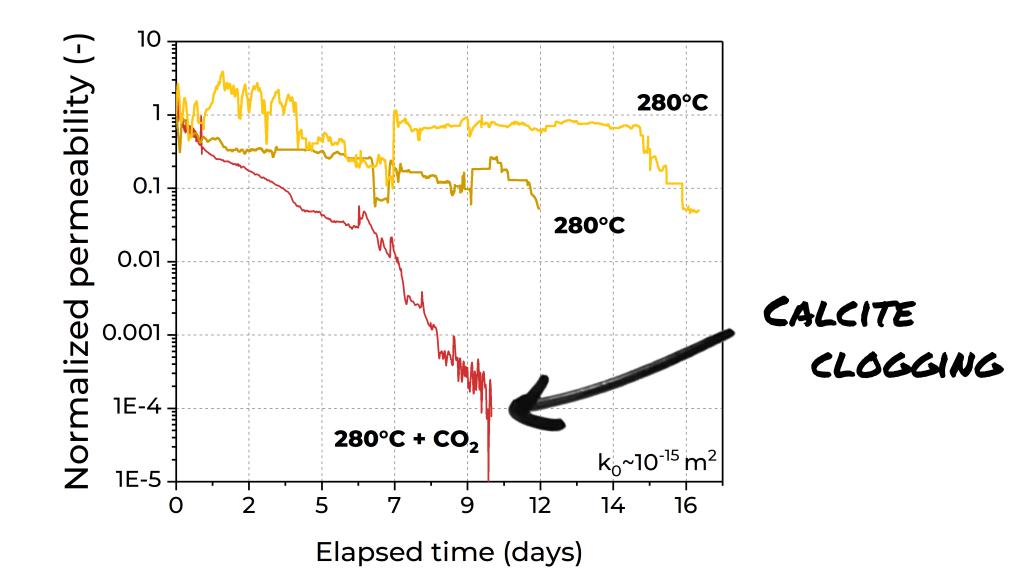
Carbonaceous matter likely generated by  $CO_2$  reduction with  $H_2$ 

 $nCO_2 + xH_2 \rightarrow C_nH_m + H_2O$ 

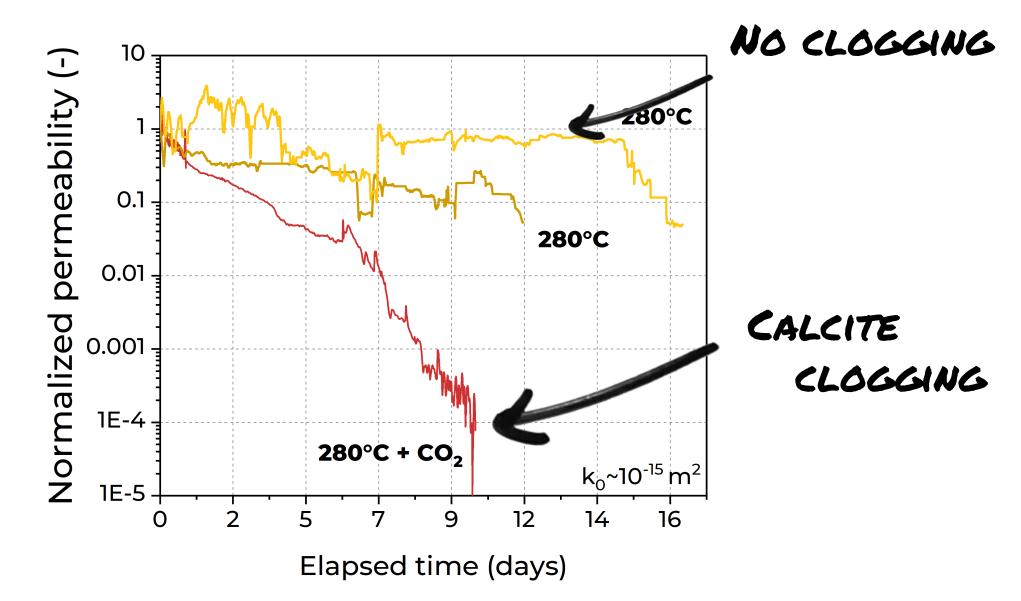
# **Permeability evolution**



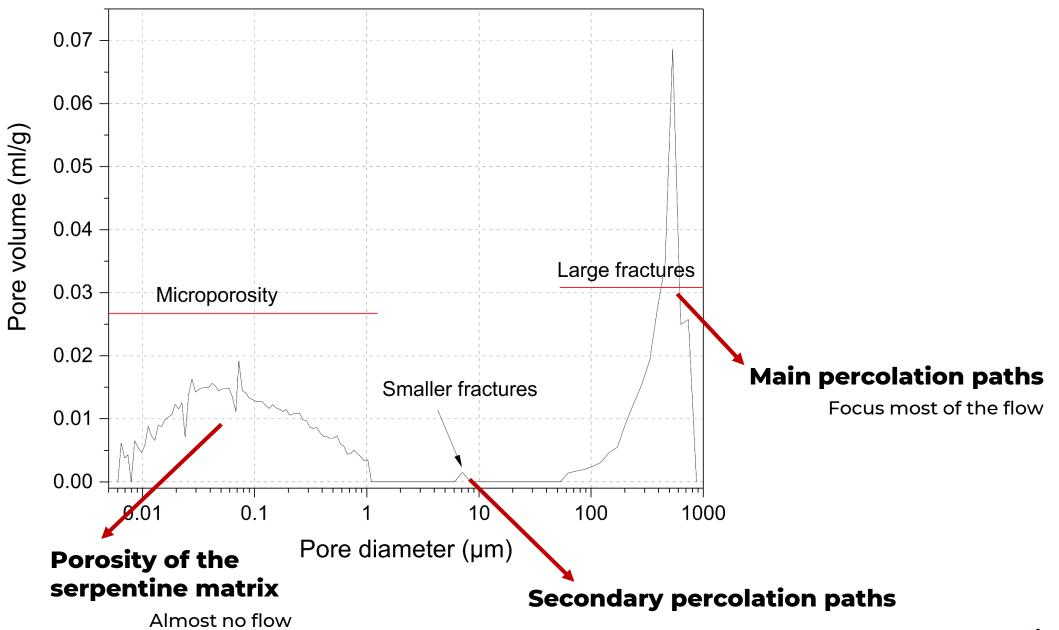
# Hydrochemical Coupling - Permeability evolution



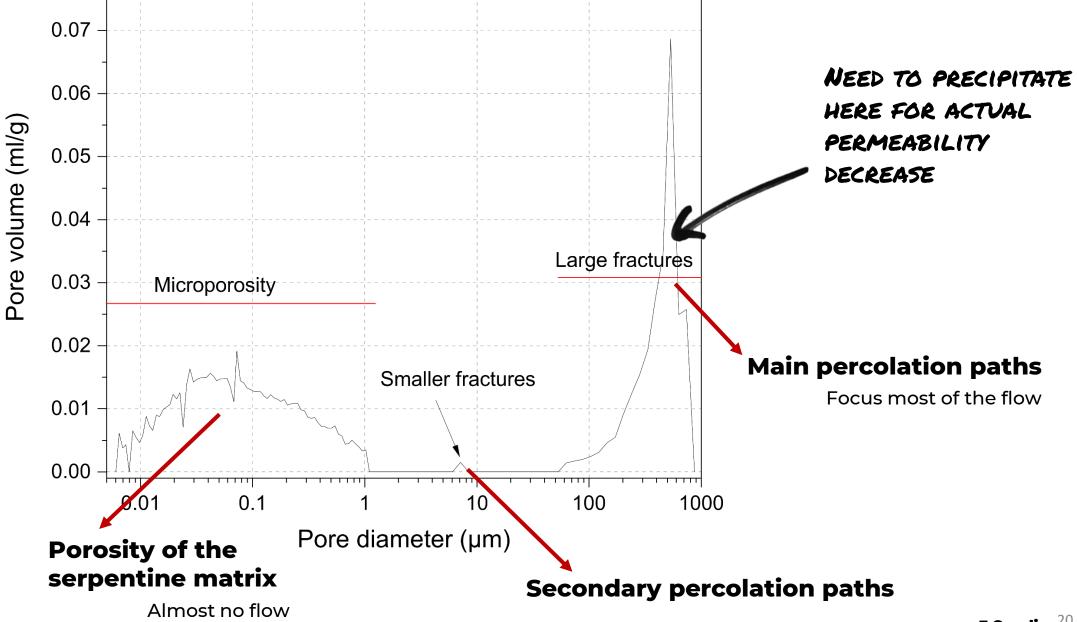
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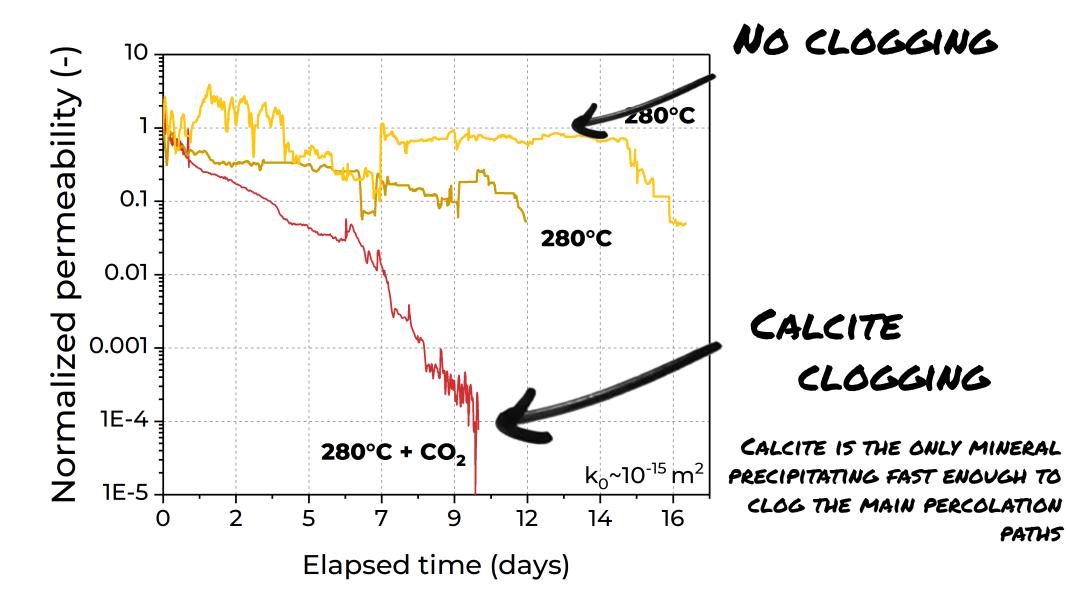
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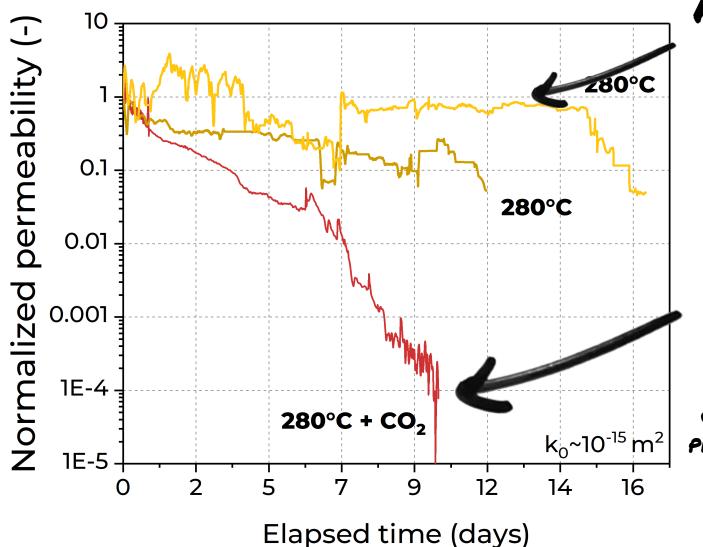
# **Hydrochemical Coupling**



# Hydrochemical Coupling - Permeability evolution



# Hydrochemical Coupling - Permeability evolution



### NO CLOGGING

NO MINERAL IS FAST ENOUGH TO PRECIPITATE IN THE MAIN PERCOLATION PATHS

CALCITE CLOGGING

CALCITE IS THE ONLY MINERAL PRECIPITATING FAST ENOUGH TO CLOG THE MAIN PERCOLATION PATHS

# **Scientific Conclusion**

### Importance of THMC coupling in Earth Science processes.

- ♦ Complicated chemistry
- ◆ Local equilibria
- ♦ Variable flow rates
- Importance of pressure and mechanics

### Oespite a large molar volume variation, silicates are not responsible for permeability drop.

It's not the variation of molar volume, it's the location of the precipitation that matters.

 $\diamond$  Despite clogging, reasonable carbonation of 5.6% of total injected CO<sub>2</sub> at 280° as well as a 3% to 8% H<sub>2</sub> yield in the CO<sub>2</sub>-free case.

## Perspectives

Potential to be a game-changer for the energy transition

♦ Market of \$1000B.

### ♦ The current state of research shows it is possible, but :

- What are the best P, T, Q, x conditions of injection?
- ♦ How does the permeability/porosity evolve during the process?
- ♦ How does the surface area evolve passivation?
- What is the influence of microbiology on the yield?

## LOTS OF SCIENCE TO DO, COLLABORATIONS ARE VERY MUCH WELCOMED!